

1.

$$Q(t) = 300e^{0.02t}$$

a. $k = 0.02$ for this problem (IN THIS CASE)

b. $Q_0 = Q(0) = 300$ because $t=0$ and $e^{0.02t} = e^0 = 1$

c.

t	0	10	20	100	1000
Q	300	366.42	447.55	2216.7	1.5×10^4

I used my graphing calculator

1. Y=
2. TBLSET
3. TABLE

3. Growth of Bacteria under ideal laboratory conditions

a. (from example Pg 392) $Q(t) = Q_0 e^{kt}$

given: $Q(0) = 100 e^{k(0)} = 100$

$$Q(20) = 100 e^{20k} = 200$$

$$e^{20k} = \frac{200}{100} = 2$$

$$\ln e^{20k} = \ln 2$$

$$20k \ln e = .6931$$

$$20k = .6931$$

$$k = \frac{.6931}{20} = \underline{\underline{.03466}}$$

Thus $Q(t) = 100 e^{.03466t}$

b.

$$100 e^{.03466t} = 1,000,000$$

$$e^{.03466t} = \frac{1,000,000}{100} = 10,000$$

$$.03466t \ln e = \ln 10^4$$

$$t = \frac{4 \ln 10}{.03466} = 265.734 = \underline{\underline{266 \text{ minutes}}}$$

c. The k for *Escherichia coli* would not change just Q_0

so $Q(t) = 1000 e^{.03466t}$

Sec 5.6 Pg 399 (#7)

2. Atmospheric Pressure (cool, I am a high mt. climber)

$$P = P_0 e^{-Kh}$$

given: $P_0 = 15 e^{-K(0)} = 15 \text{ lb/in}^2 \text{ at sea level}$

$$P(4000) = 15 e^{-4000K} = 12.5$$

$$e^{-4000K} = \frac{12.5}{15} = .833$$

$$\ln e^{-4000K} = \ln(.833)$$

$$-4000K = -.182322$$

$$K = \frac{-.182322}{-4000} = .00004558$$

$$P(12000) = 15 e^{(.00004558)(12000)}$$

$$P(12000) = 15 e^{-.546966}$$

$$P(12000) = 8.680544 = 8.7 \text{ lb/in}^2 \text{ at } 12,000 \text{ ft}$$

$$P'(h) = \frac{d}{dh} (15 e^{-0.00004558h})$$

$$P'(h) = -0.0006837 e^{-0.00004558h}$$

$$P'(12000) = -0.0006837 e^{-0.00004558(12,000)}$$

$$P'(12000) = -0.00039566157$$

Thus the atmospheric pressure is declining at the rate of about $0.0004 \text{ lb/in}^2/\text{ft}$

The highest mountain in the world is Aconcagua (according to Geography prof.)

at 22,841 ft

$$P(22,841) = 15 e^{(-.00004558)(22,841)} = 5.296 \text{ lb/in}^2 \text{ at where I was}$$

$$P'(22,841) = -0.0006837 e^{-0.00004558(22,841)} = -0.00024139 \text{ lb/in}^2/\text{ft}$$

extra because I was interested

Sec. 5.6 cont.

Pg 399 (8 ~~cont.~~, 9)

8. Given: $Q(t) = Q_0 \cdot 2^{-t/140}$

$Q(280) = 20 \text{ mg}$

$Q(280) = Q_0 \cdot 2^{-280/140} = 20$

$Q_0 \cdot 2^{-2} = 20$

$\frac{1}{4} Q_0 = 20$ Thus $Q_0 = 80 \text{ mg}$

Note [from examples 2 & 3] $\left[\begin{matrix} P_3 393 & P_3 394 \\ 2 & 3 \end{matrix} \right]$ decay constant $\downarrow -k$

9. P-32 at T is given by $Q(T) = Q_0 e^{-kT}$

First find k for P-32

$$\begin{cases} \frac{1}{2} Q_0 = Q_0 e^{-14.2k} \\ e^{-14.2k} = \frac{1}{2} \\ -14.2k = \ln \frac{1}{2} \\ k = -\frac{\ln(1/2)}{14.2} \approx 0.0488 \end{cases}$$

So $Q(T) = 100 e^{-0.0488T}$

$Q(7.1) = 100 e^{-0.0488(7.1)}$
 $\approx 100 e^{-0.34648}$

$Q(7.1) \approx 70.717 \text{ grams}$

$Q'(T) = 100 (-0.0488) e^{-0.0488T} = -4.88 e^{-0.0488T}$

$Q'(7.1) = -4.88 e^{-0.0488(7.1)} \approx -3.451 \text{ g/day}$

rate of decay $\approx -3.451 \text{ g/day}$

$-0.0488(7.1) = -4.88e$

$-0.0488T$

In case you cannot read last two lines rewritten